

Badische Landesbibliothek Karlsruhe

Digitale Sammlung der Badischen Landesbibliothek Karlsruhe

The young man's book of amusement

Halifax, 1848

To Bleach Prints and Printed Books

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tint gradually fades and becomes brown, and the redness of this brown tint gradually increases till the mineral has recovered its primitive brilliant red colour. A green ruby suffered no change of colour from heat, and a bluish green sapphire became much paler at a high heat, but resumed its original colour by cooling.

To Bleach Prints and Printed Books.

Simple immersion in oxygenated muriatic acid, letting the article remain in it, a longer or shorter space of time, according to the strength of the liquor, will be sufficient to whiten an engraving: if it be required to whiten the paper of a bound book, as it is necessary that all the leaves should be moistened by the acid, care must be taken to open the book well, and to make the boards rest on the edge of the vessel, in such a manner that the paper alone shall be dipped in the liquid; the leaves must be separated from each other, in order that they may be equally moistened on both sides. The liquor assumes a yellow tint, and the paper becomes white in the same proportion; at the end of two or three hours, the book may be taken from the acid liquor, and plunged into pure water with the same care and precaution as recommended in regard to the acid liquor, that the water may touch both sides of each leaf. The water must be renewed every hour, to extract the acid remaining in the paper, and to dissipate the disagreeable smell.

To Remove

The stains of ink, removed by all acids, which are leaved, which are leaved, the stained substance with five or six times diluted to the spot, &c., repeating the process as necessary. In the acids. A solution of tartaric (oxalic) or tartaric acid is the most efficacious; and repeating them; and repeating them in cleaning the writing on the paper.

Dr. Woll

The cold produce in circumstances, very which readily evaporating that process, and similar acids they contain evaporation of water washed receiver, which its vapour,